

# EC 3210 Solutions

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## Assignment 5

4.1. Given a coefficient  $B_{ij}$  of  $10^{19} \text{ m}^3 \cdot \text{W}^{-1} \cdot \text{s}^{-3}$  for a 650 nm transition in a material with  $n = 1$ ,

- Calculate  $A_{ji}$  and the spontaneous lifetime of the transition.
- Calculate the ratio of  $A_{ij}$  to the stimulated transition rate if the irradiance is  $10 \text{ mW/mm}^2$  concentrated at the transition wavelength and  $g(\nu_0) = 10^{-10}$ .

a. The desired quantities are

$$A_{ij} = \frac{8\pi n^3 h}{\lambda^3} B_{ij} = \left( \frac{8\pi(1)^3 6.63 \times 10^{-34}}{(650 \times 10^{-9})^3} \right) (10^{19}) = 6.06 \times 10^5 \text{ s}^{-1} \quad (1)$$

and

$$\tau_s = \frac{1}{A_{ij}} = \frac{1}{6.06 \times 10^5} = 1.649 \times 10^{-6} \text{ s}. \quad (2)$$

b. We note that  $I = 10 \times 10^{-3} \text{ W} \cdot \text{mm}^{-2}$  is equal to  $10 \times 10^3 \text{ W} \cdot \text{m}^{-2}$ , and, so,

$$\begin{aligned} \frac{A_{ij}}{B_{ij} \rho_\nu} &= \frac{A_{ij}}{B_{ij} \left( \frac{nI \delta(\nu - \nu_0)}{c} \right) g(\nu)} = \frac{A_{ij} c}{B_{ij} n I (\nu_0) g(\nu_0)} \\ &= \frac{(6.06 \times 10^5)(3.0 \times 10^8)}{(1 \times 10^{19})(1)(1 \times 10^4)(1 \times 10^{-10})} = 18.17. \end{aligned} \quad (3)$$

4.2 The loss coefficient of a material is 1% per mm.

- Calculate the fraction of the power transmitted through a 5 cm thickness of the material.
- Find the value of the loss coefficient  $\alpha$ .

a. We find

$$\alpha = \frac{0.01}{0.001} = 10 \text{ m}^{-1} \quad (4)$$

$$\frac{I_{\text{out}}}{I_{\text{in}}} = e^{-\alpha D} = e^{-(10)(5 \times 10^{-2})} = 0.607 = 60.7\%. \quad (5)$$

b. We have already found this value. From above,  $\alpha = 10 \text{ m}^{-1}$ .

4.3 A material exhibits a gain of 1% per meter when pumped at a certain power level.

- Calculate the gain coefficient  $\beta$ .
  - Find the length of material required to double the power in a wave.
- a. We find that, for 1 meter,

$$\beta \approx \frac{\alpha L}{L} = \frac{(0.01)(1)}{1} = 0.01 \text{ m}^{-1}. \quad (6)$$

b. To double the power requires

$$G = e^{+\beta z} = 2, \quad (7a)$$

so

$$\ln 2 = \beta z \quad (7b)$$

and

$$z = \frac{\ln 2}{\beta} = \frac{\ln 2}{0.01} = 69.3 \text{ m}. \quad (7c)$$

**4.4** A ruby laser operates at 694 nm from an upper level to the ground state.

a. Calculate the temperature that is required to place 50% of the total number of atoms in the upper state (leaving 50% in the ground state).

b. Calculate the temperature that is required to place 10% of the total number of atoms in the upper state (leaving 90% in the ground state).

c. Calculate the fraction of the total population that are in the upper state when T is 300K.

See Fig. 1 for the population levels.

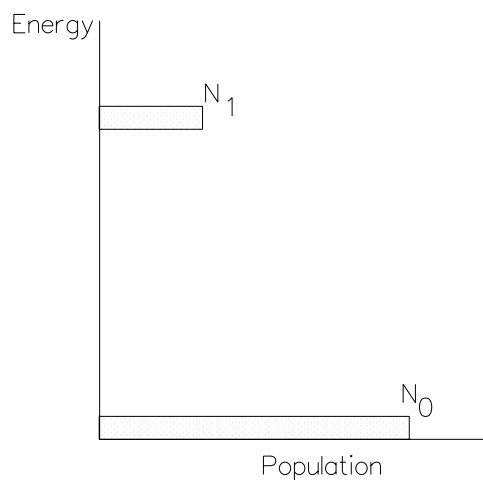


Figure 1: Energy levels and populations for Problem 4.4

a. The temperature required to place 50% of the total atoms in the upper state (leaving 50% in the lower state) is found from

$$\frac{N_1}{N_0} = \exp\left(-\frac{\Delta E}{kT}\right) = 1, \quad (8a)$$

so

$$-\frac{\Delta E}{kT} = 0 \quad (8b)$$

and

$$T \rightarrow \infty. \quad (8c)$$

It would take an infinite temperature to get 50% of the atoms into the upper level *if the material were in thermal equilibrium*.

b. To place 10% in the upper level would require

$$\frac{N_1}{N_0} = \exp\left(-\frac{\Delta E}{kT}\right) = \frac{0.1}{0.9}, \quad (9a)$$

so

$$-\frac{\Delta E}{kT} = \ln\left(\frac{1}{9}\right) = -2.20 \quad (9b)$$

and

$$T = -\frac{\Delta E}{k(-2.20)} = \frac{hc}{\lambda k(2.20)} = 9.42 \times 10^3 \text{ K}. \quad (9c)$$

c. The fraction of the *total* population that is in the upper stage when  $T = 300 \text{ K}$  is

$$\frac{N_1}{N_{\text{total}}} = \frac{N_1}{N_0 + N_1} = \frac{1}{1 + \frac{N_0}{N_1}}, \quad (10a)$$

where

$$\begin{aligned} \frac{N_0}{N_1} &= \frac{1}{\frac{N_1}{N_0}} = \exp\left(\frac{\Delta E}{kT}\right) = \exp\left(\frac{hc}{\lambda kT}\right) \\ &= \exp\left(\frac{(6.63 \times 10^{-34})(3.0 \times 10^8)}{(694 \times 10^{-9})(1.38 \times 10^{-23})(300)}\right) = 1.161 \times 10^{30} \end{aligned} \quad (10b)$$

and, so,

$$\frac{N_1}{N_{\text{total}}} = \frac{1}{1 + \frac{N_0}{N_1}} = \frac{1}{1 + 1.161 \times 10^{30}} = 8.61 \times 10^{-31}. \quad (10c)$$

Hence, we find that the upper level is *really* empty at thermal equilibrium at room temperature.

**4.5** Calculate the population inversion required per unit volume to give a small-signal gain coefficient of  $0.5 \text{ m}^{-1}$  in a  $\text{CO}_2$  laser. Assume that  $A_{ij}$  is  $200 \text{ s}^{-1}$ ,  $g(\nu_0) \approx 1/\Delta\nu = 1 \times 10^{-9} \text{ s}$ , and  $n = 1$ .

First we note that

$$\beta = \frac{(N_j - N_i)B_{ij}nh\nu_0g(\nu_o)}{c\text{Vol}}, \quad (11)$$

so

$$\frac{N_j - N_i}{\text{Vol}} = \frac{\lambda\beta}{B_{ij}nhg(\nu_0)}. \quad (12)$$

We also note that

$$B_{ij} = \frac{\lambda^3 A_{ij}}{8\pi n^3 h} = \frac{(10.6 \times 10^{-3})(200)}{8\pi(1)^3(6.63 \times 10^{-34})} = 1.43 \times 10^{19} \quad (13)$$

and, so,

$$\begin{aligned} \frac{N_j - N_i}{\text{Vol}} &= \frac{(10.6 \times 10^{-6})(0.5)}{(1.43 \times 10^{19})(1)(6.64 \times 10^{-34})(1 \times 10^{-9})} \\ &= 5.59 \times 10^{17} \text{ molecules} \cdot \text{m}^{-3}. \end{aligned} \quad (14)$$